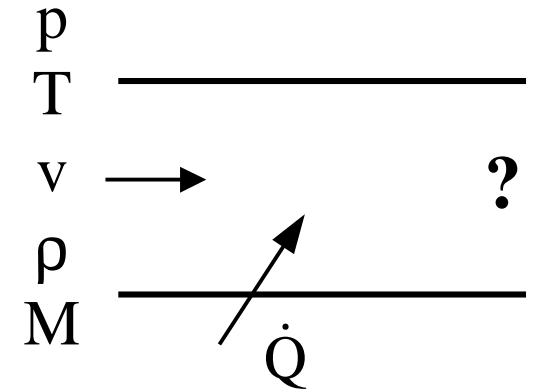


Rayleigh Flow - Thermodynamics

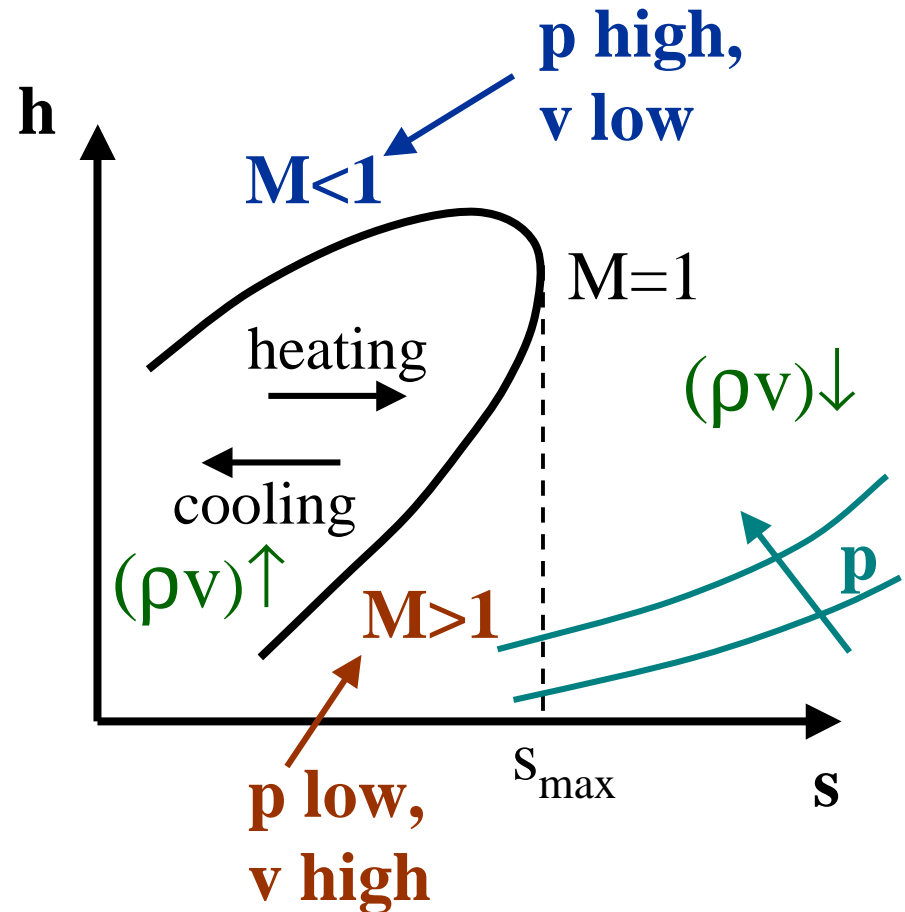
- Steady, 1-d, constant area, inviscid flow with no external work but *with reversible heat transfer (heating or cooling)*



- Conserved quantities (mass, momentum eqs.)
 - since $A=\text{const}$: mass flux= $\rho v=\text{constant}\equiv G$
 - since no forces but pressure: $p+\rho v^2=\text{constant}$
 - combining: $p+G^2/\rho=\text{constant}$
- On h - s diagram, can draw this **Rayleigh Line**

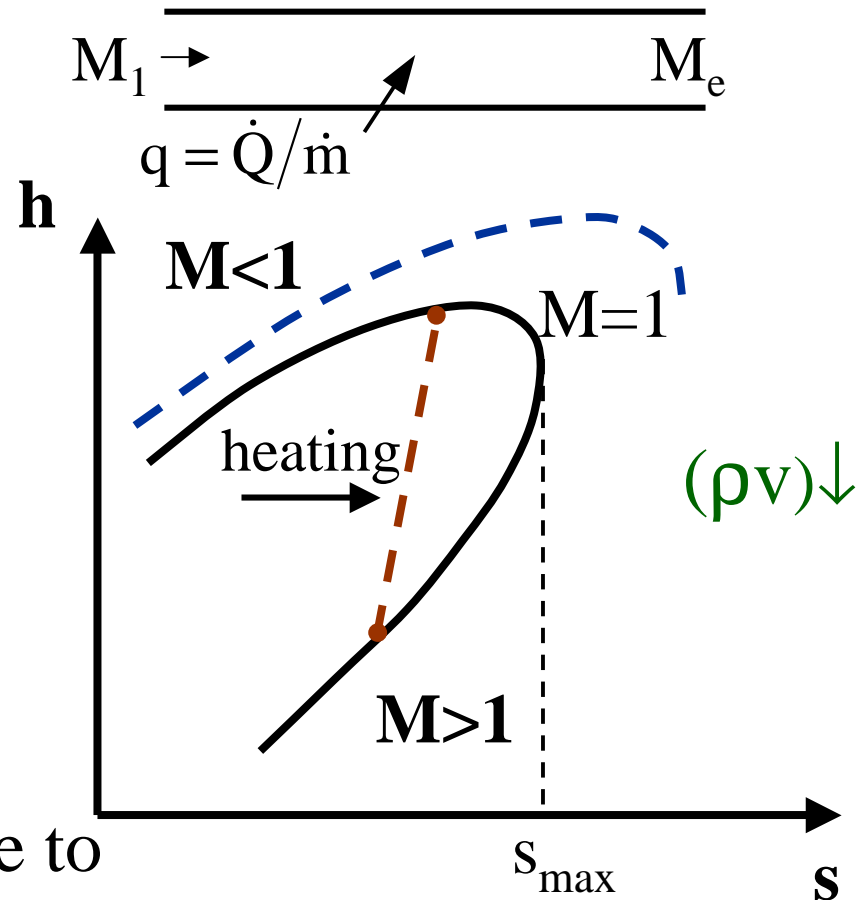
Rayleigh Line

- $p + \rho v^2 = \text{constant}$,
 \Rightarrow high p results in low v
- $ds = \delta Q / T$ (reversible)
 - heating increases s
 - cooling decreases s
 - $M=1$ at maximum s
- Change mass flux
 new Rayleigh line



Rayleigh Line - Choking

- Heat addition can only increase entropy
- For **enough heating**, $M_e=1$ ($Q_{in}=Q_{max}$)
 - heat addition can lead to choked flow
- What happens if $Q_{in} > Q_{max}$
 - **subsonic flow**: must move to different Rayleigh line (---), i.e., lower mass flux
 - **supersonic flow**: get a shock (---)



stays on same Rayleigh line: ρv and $p + \rho v^2$ also const. for normal shock

Differential Mach Equations

- Simplify (X.4-5) for $f=dA=0$

$$\frac{dM^2}{M^2} = \frac{\left(1 + \gamma M^2\right) \left(1 + \frac{\gamma - 1}{2} M^2\right) \delta q}{\left(1 - M^2\right) c_p T_o} \quad (\text{X.21})$$

$$\frac{dp}{p} = \frac{-\gamma M^2}{1 + \gamma M^2} \frac{dM^2}{M^2} \quad (\text{X.22})$$

$$\frac{dT}{T} = \frac{dh}{h} = \frac{1 - \gamma M^2}{1 + \gamma M^2} \frac{dM^2}{M^2} \quad (\text{X.23})$$

$$\frac{dv}{v} = -\frac{dp}{\rho} = \frac{-1}{1 + \gamma M^2} \frac{dM^2}{M^2} \quad (\text{X.24})$$

$$\frac{dT_o}{T_o} = \frac{\delta q}{c_p T_o} \quad (\text{X.25})$$

$$\frac{dp_o}{p_o} = -\frac{\gamma}{2} M^2 \frac{dT_o}{T_o} \quad (\text{X.26})$$

$$\frac{ds}{c_p} = \frac{1 - M^2}{1 + \gamma M^2} \frac{dM^2}{M^2} \quad (\text{X.27})$$

- sign changes**

Property Variations

- What can change sign in previous equations
 - $(1-M^2)$ and δq
 - $(1-\gamma M^2)$ in dh, dT for $M < 1$

- s, T_o just like q
- p_o opposite of T_o and s
- Heating: $M \rightarrow 1$; cooling opposite
- p, ρ opposite of v ($p + \rho v^2 = \text{const}$)
- h, T same as p, ρ unless $M^2 < 1/\gamma$
(low speed flow: T oppos. p, ρ)

δq	$M < 1$		$M > 1$	
	< 0	> 0	< 0	> 0
s, T_o	↓	↑	↓	↑
p_o	↑	↓	↑	↓
M, v	↓	↑	↑	↓
p, ρ	↑	↓	↓	↑
h, T			↓	↑
$M^2 < \gamma^{-1}$	↓	↑		
$M^2 > \gamma^{-1}$	↑	↓		

$\delta q < 0 \Rightarrow$ cooling
 $\delta q > 0 \Rightarrow$ heating

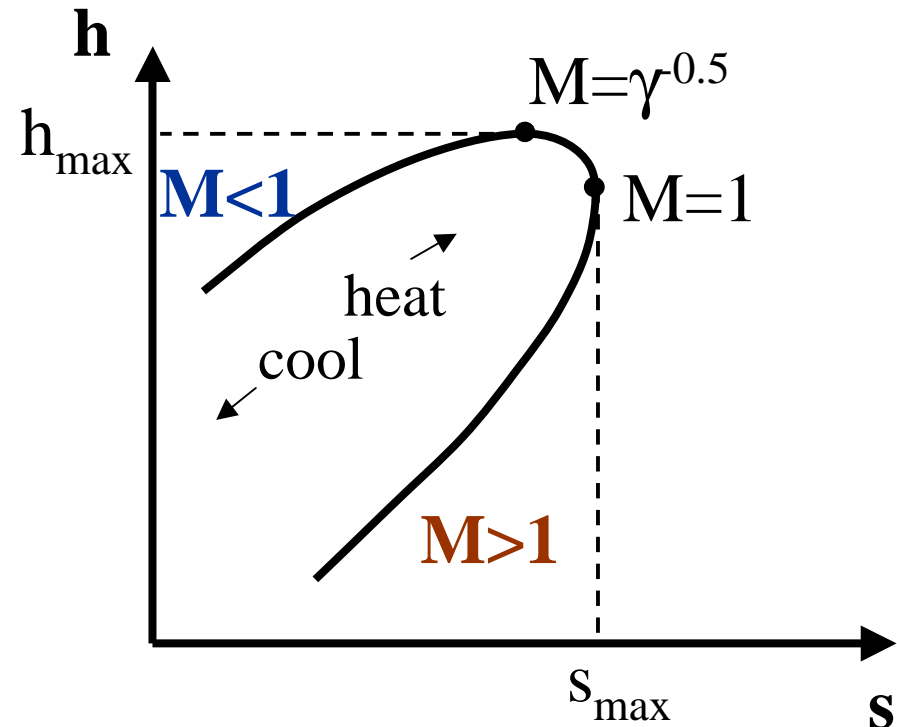
Rayleigh Line Extremum

- Combine X.23 and X.27

$$\frac{dh}{ds} = \frac{h}{c_p} \frac{1 - \gamma M^2}{1 - M^2}$$

$$\frac{dh}{ds} = 0 \text{ for } M = \frac{1}{\sqrt{\gamma}}$$

$$\frac{dh}{ds} = \infty \text{ for } M = 1$$



Integrated Mach Equations

- Integrating (X.22-27)

$$\frac{dT_o}{T_o} = \frac{\delta q}{c_p T_o} \Rightarrow \int dT_o = \int \frac{\delta q}{c_p} \Rightarrow (T_{o2} - T_{o1}) = \frac{q_{12}}{c_p} \quad \text{(X.28)}$$

(energy) **(X.32)**

$$\frac{p_2}{p_1} = \frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \quad \text{(X.29)}$$

$$\text{(X.30)}$$

$$\frac{T_2}{T_1} = \left(\frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)^2 \left(\frac{M_2}{M_1} \right)^2$$

$$\frac{p_{o2}}{p_{o1}} = \frac{p_2 \frac{p_{o2}}{p_2}}{p_1 \frac{p_{o1}}{p_1}} = \frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \left(\frac{1 + \frac{\gamma-1}{2} M_2^2}{1 + \frac{\gamma-1}{2} M_1^2} \right)^{\frac{\gamma}{\gamma-1}}$$

$$\frac{T_{o2}}{T_{o1}} = \left(\frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)^2 \left(\frac{M_2}{M_1} \right)^2 \left(\frac{1 + \frac{\gamma-1}{2} M_2^2}{1 + \frac{\gamma-1}{2} M_1^2} \right)^2 \quad \text{(X.33)}$$

$$\frac{v_1}{v_2} = \frac{\rho_2}{\rho_1} = \frac{1 + \gamma M_2^2}{1 + \gamma M_1^2} \left(\frac{M_1}{M_2} \right)^2 \quad \text{(X.31)}$$

$$\frac{s_2 - s_1}{c_p} = \ln \left[\left(\frac{M_2}{M_1} \right)^2 \left(\frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)^{(\gamma+1)/\gamma} \right] \quad \text{(X.34)}$$

Solution Approaches

- Two methods to “solve” Rayleigh problems, i.e., for state change from 1→2
- In both, need to “know” one state (including M) and something about other state (1 TD property or q or M)
- **Method 1:** direct solution of integrated equations (X.28-34)
 - requires iterative solution
- **Method 2:** tabular solutions using reference condition
 - uses sonic reference condition (similar to A/A^* and Fanno methods)
 - sonic condition based on heat transfer required for flow to go sonic; $M_2=1, T_{o2}=T_o^*$

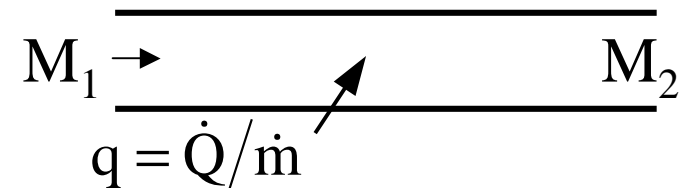
Direct Solution

- Use two equations to find M change
 - e.g., if state 1 and q_{12} known use X.28 and X.33

a) get T_o ratio from X.28

$$\frac{T_{o2}}{T_{o1}} = 1 + \frac{q_{12}}{c_p T_{o1}}$$

doesn't matter how q changes along length, only total q required



b) get M_2 from X.33 (must know M_1)

$$\left(\frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)^2 \left(\frac{M_2}{M_1} \right)^2 \left(\frac{1 + \frac{\gamma-1}{2} M_2^2}{1 + \frac{\gamma-1}{2} M_1^2} \right) = \frac{T_{o2}}{T_{o1}} \text{ requires iterative solution for } M_2$$

c) get rest of property changes M_2 from X.29-34, e.g., from X.29

$$\frac{p_2}{p_1} = \frac{1 + \gamma M_1^2}{1 + \gamma M_2^2}$$

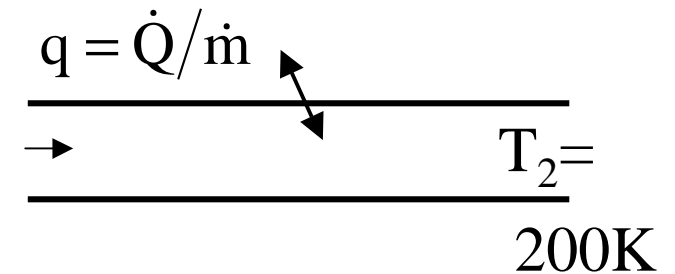
Cooled Duct Example

- **Given:** Nitrogen flowing in constant area duct, enters at $M=3$, $T=250$ K, $p=50$ kPa and exits with temperature of 200 K

- **Find:**

1. M_2
2. q (including direction)

$$\begin{aligned}
 M_1 &= 3.0 \\
 T_1 &= 250 \text{ K} \\
 p_1 &= 50 \text{ kPa}
 \end{aligned}$$



- **Assume:** N_2 is tpg/cpg, $\gamma=1.4$, steady, reversible, no work

Cooled Duct - Solution

- Analysis: (from X.30)

$$M_1=3.0 \quad q = \dot{Q}/\dot{m} \quad T_2=200K$$

$$T_1=250K \quad \rightarrow$$

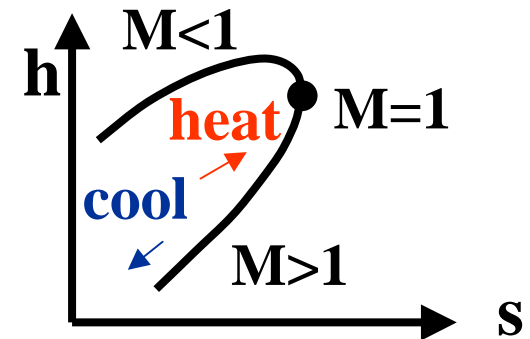
$$p_1=50 \text{ kPa}$$

$$- M_2 \left(\frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)^2 \left(\frac{M_2}{M_1} \right)^2 = \frac{T_2}{T_1}$$

$$\left(\frac{1 + 1.4(3)^2}{1 + 1.4M_2^2} \right)^2 \left(\frac{M_2}{3} \right)^2 = \frac{200}{250} = 0.80$$

$$\Rightarrow M_2 = 0.21, 3.41$$

since started $M > 1$, can't be subsonic



- q (energy: ins = outs)

$$q = c_p (T_{o1} - T_{o2})$$

$$= \frac{1.4}{1.4 - 1} \frac{8314 \text{ J/kmolK}}{28 \text{ kg/kmol}} (700 - 665) \text{ K}$$

$$= 36 \text{ kJ/kg}$$

$$T_{o1} = T_1 \left(1 + \frac{\gamma - 1}{2} M_1^2 \right)$$

$$= 250 \text{ K} \left(1 + \frac{1.4 - 1}{2} 9 \right) = 700 \text{ K}$$

$$T_{o2} = 200 \text{ K} \left(1 + 0.2(3.41)^2 \right) = 665 \text{ K}$$

Reference Solution Method

- Reference state has $M=1$, $T \equiv T^*$, $p \equiv p^*$, ... \Rightarrow sonic, but not same state as A^* or Fanno
- State 2 \rightarrow sonic in integrated equations (X.29-34)

$$\frac{p}{p^*} = \frac{1 + \gamma}{1 + \gamma M^2} \quad \text{(X.35)}$$

$$\frac{T}{T^*} = M^2 \left(\frac{1 + \gamma}{1 + \gamma M^2} \right)^2 \quad \text{(X.36)}$$

$$\frac{v}{v^*} = \frac{\rho^*}{\rho} = \frac{1 + \gamma}{1 + \gamma M^2} M^2 \quad \text{(X.37)}$$

- Note:** $prop/prop^* = f(M)$ only
- Values for $\gamma=1.4$ in Appendix F, John

$$\text{(X.38)} \quad \frac{p_o}{p_o^*} = \frac{1 + \gamma}{1 + \gamma M^2} \left(\frac{1 + \frac{\gamma - 1}{2} M^2}{1 + \frac{\gamma - 1}{2}} \right)^{\frac{\gamma}{\gamma - 1}}$$

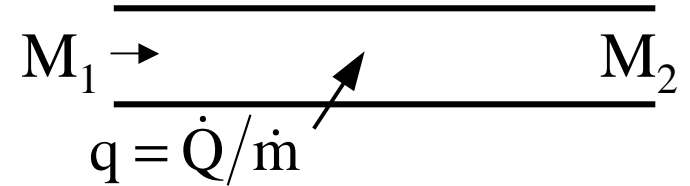
$$\text{(X.39)} \quad \frac{T_o}{T_o^*} = \left(\frac{1 + \gamma}{1 + \gamma M^2} \right)^2 M^2 \frac{1 + \frac{\gamma - 1}{2} M^2}{1 + \frac{\gamma - 1}{2}}$$

$$\text{(X.40)} \quad \frac{s - s^*}{c_p} = \ln \left\{ M^2 \left(\frac{1 + \gamma}{1 + \gamma M^2} \right)^{\frac{1 + \gamma}{\gamma}} \right\}$$

Use of Tables

- To get change in M , use change in property ratio (like using A/A^*), e.g.,

$$\frac{T_{o2}}{T_{o1}} = \frac{T_{o2}/T_o^*}{T_{o1}/T_o^*} = \frac{T_o/T_o^*|_{M_2}}{T_o/T_o^*|_{M_1}}$$



- For example, if state 1 and q_{12} known

1) again get T_o ratio from X.28

$$\frac{T_{o2}}{T_{o1}} = 1 + \frac{q_{12}}{c_p T_{o1}}$$

2) look up T_o/T_o^* at M_1

3) calculate T_o/T_o^* at M_2

$$\frac{T_{o2}}{T_o^*} = \frac{T_{o2}}{T_{o1}} \times \frac{T_{o1}}{T_o^*}$$

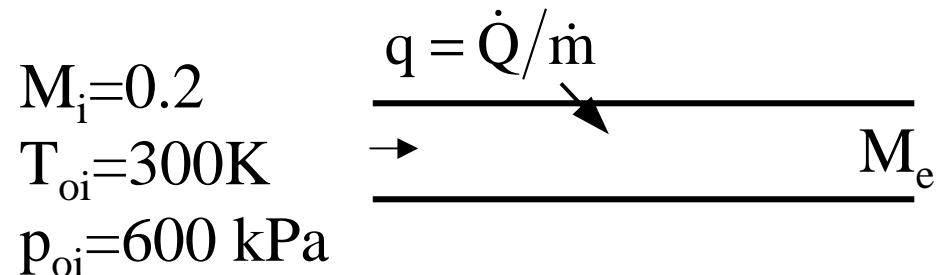
4) look up M_2 that corresponds to calculated T_o/T_o^*

Heated Duct Example

- **Given:** Air flowing in constant area duct, enters at $M=0.2$ and given inlet conditions and a heating rate of 765 kJ/kg

- **Find:**

1. M_e, p_{oe}, T_e
2. q_{\max} for $M_e=1$

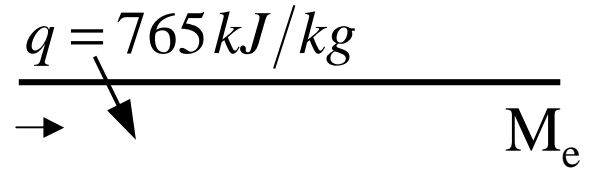


- **Assume:** air is tpg/cpg, $\gamma=1.4$, steady, reversible, no work

Heated Duct - Solution

• **Analysis:**

$M_i = 0.2$
 $T_{oi} = 300\text{K}$
 $p_{oi} = 600\text{ kPa}$



- M_e
 (X.28, energy)

$$\frac{T_{oe}}{T_{oi}} = 1 + \frac{q}{c_p T_{oi}}$$

$$= 1 + \frac{765 \times 10^3 \text{ J/kg}}{1004 \frac{\text{J}}{\text{kgK}} (300\text{K})} = 3.54$$

M	To/To*	po/po*
0.19	0.15814	1.23765
0.20	0.17355	1.23460
0.21	0.18943	1.23142
0.44	0.59748	1.13936
0.45	0.61393	1.13508
0.46	0.63007	1.13082
3.53	0.61393	5.47054

$$\frac{T_{oe}}{T_o^*} = \frac{T_{oe}}{T_{oi}} \frac{T_{oi}}{T_o^*} \Big|_{M=0.2} = 3.54(0.17355) = 0.614 \Rightarrow M_e = 0.45$$

(Appendix F)

another solution for $M=3.53$, but since started with $M < 1$, can't be supersonic

- P_{oe}

$$p_{oe} = \frac{p_{oe}^*}{p_o^*} \frac{p_o^*}{p_{oi}} p_{oi} = 1.1351 \frac{1}{1.2346} 600\text{kPa} = 552\text{kPa}$$

Heated Duct Solution (con't)

– T_e

$$T_e = \frac{T_e}{T_{oe}} \frac{T_{oe}}{T_{oi}} T_{oi}$$

$$\begin{aligned} M_i &= 0.2 \\ T_{oi} &= 300\text{K} \\ p_{oi} &= 600\text{ kPa} \end{aligned}$$

$$\frac{q = 765\text{ kJ/kg}}{\longrightarrow \quad \quad \quad \longrightarrow} M_e$$

$$T_e = \frac{1}{1 + \frac{0.4}{2} 0.45^2} 3.54(300\text{K}) = 1021\text{K}$$

– q_{max}

requires choked exit, $T_{oe} = T^*$

$$q_{max} = c_p (T_o^* - T_o) = c_p T_{o1} \left(\frac{T_o^*}{T_{o1}} - 1 \right)$$

$$T_o^* = \frac{300\text{K}}{0.17355} = 1728\text{K}$$

$$T^* = T_o^* / \left(1 + \frac{\gamma-1}{2} \right) = 1440\text{K}$$

$$= 1004 \frac{\text{J}}{\text{kgK}} 300\text{K} \left(\frac{1}{0.17355} - 1 \right) = 1.43 \frac{\text{MJ}}{\text{kg}}$$

Combustor Example

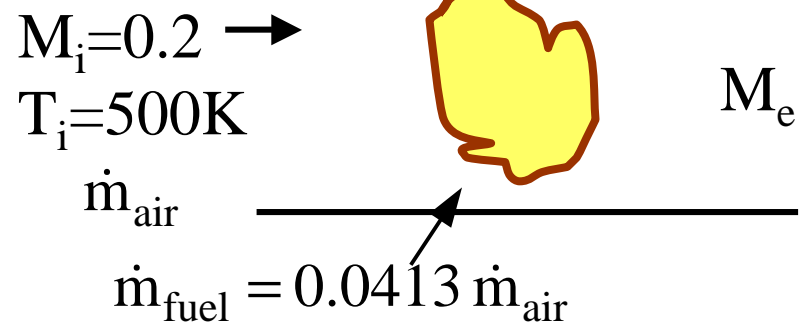
- **Given:** Air entering constant combustor, enters at $M=0.2$ and given inlet conditions.

$$q = \text{heating value of fuel (45.5 MJ/kg}_{\text{fuel}}) \times \frac{\dot{m}_{\text{fuel}}}{\dot{m}_{\text{air}}}$$

- **Find:**

1. T_e and compare to value you would calculate if neglected kinetic energy change

2. p_o loss



- **Assume:** air is tpg/cpg, $\gamma=1.4$, steady, reversible, no work,
 $\frac{\dot{m}_{\text{fuel}}}{\dot{m}_{\text{air}}} \ll 1$ (can neglect mass addition, just use q)

Combustor Solution

• **Analysis:** (energy)

$$- T_e \quad \frac{T_{oe}}{T_{oi}} = 1 + \frac{q}{c_p T_{oi}}$$

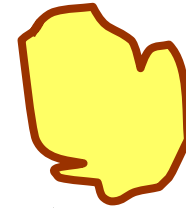
? combustion just another kind of heat addition

$$M_i = 0.2 \rightarrow$$

$$T_i = 500K$$

$$\dot{m}_{air}$$

$$\dot{m}_{fuel} = 0.0413 \dot{m}_{air}$$



M_e

$$q = 0.0413 \frac{\text{kg}_{fuel}}{\text{kg}_{air}} \left(45.5 \frac{\text{MJ}}{\text{kg}_{fuel}} \right) = 1.88 \frac{\text{MJ}}{\text{kg}_{air}}$$

$$T_{oi} = T_i \left(1 + \frac{\gamma - 1}{2} M^2 \right) = 500K(1.008) = 504K$$

$$T_{oe} = T_{oi} \left(1 + \frac{1.88 \times 10^6}{1004(504)} \right) = 504K(4.715) = 2376K$$

even for this subsonic flow, kinetic energy reduces final T by 160K

$$\frac{T_{oe}}{T_o^*} = \frac{T_{oe}}{T_{oi}} \frac{T_{oi}}{T_o^*} = 4.715(0.1736) = 0.818 \Rightarrow M_e = 0.60$$

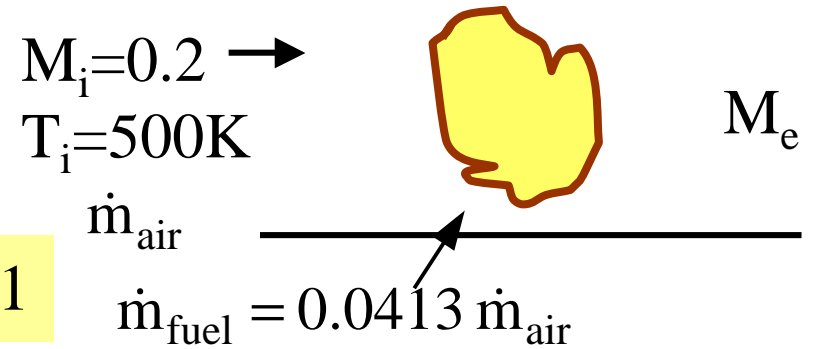
$$T_e = \frac{T_{oe}}{1 + M^2(\gamma - 1)/2} = \frac{2376K}{1 + 0.2(0.6)^2} = 2216K$$

Combustor Solution (con't)

– p_{oe}/p_{oi}

$$\frac{p_{oe}}{p_{oi}} = \frac{p_{oe}}{p_o^*} \frac{p_o^*}{p_{oi}} = 1.0753 \frac{1}{1.2346} = 0.871$$


 (Appendix F)



- So heat addition (burning fuel) in an engine gives p_o loss as was case with friction
- As we lose p_o (~13% loss here)
 - less thrust
 - less work output

Choking

- Heating of a flow can lead to choking just like friction
- For supersonic flow, depending on back pressure, can get
 - underexpanded flow
 - overexpanded flow
 - shock inside duct

