

Stagnation Properties and Mach Number

- Rewrite stagnation properties in terms of Mach number for thermally and calorically perfect gases

- **Stagnation Temperature**

- from energy conservation:
no work but flow work and **adiabatic**

⇒ T_o (and h_o) **constant** for **adiabatic flow**

$$T_o = T + \frac{1}{c_p} \frac{v^2}{2} = T + \frac{\gamma - 1}{2} \frac{v^2}{\gamma R}$$

$$\frac{T_o}{T} = 1 + \frac{\gamma - 1}{2} \frac{v^2}{\gamma R T} = 1 + \frac{\gamma - 1}{2} \frac{v^2}{a^2}$$

$$\frac{T_o}{T} = 1 + \frac{\gamma - 1}{2} M^2 \quad \text{(VI.6)}$$

- **Stagnation Pressure**

- from entropy conservation:
reversible and adiabatic
⇒ **isentropic** ($\Delta s = 0$)

⇒ p_o (and s_o) **constant** if **also reversible**

$$\frac{p_o}{p} = \left(\frac{T_o}{T} \right)^{\frac{\gamma}{\gamma - 1}} \quad \text{from state eq. for isen. process}$$

$$\frac{p_o}{p} = \left(1 + \frac{\gamma - 1}{2} M^2 \right)^{\frac{\gamma}{\gamma - 1}} \quad \text{(VI.7)}$$

Compressible p_o and Bernoulli Equation

- Incompressible flow, **Bernoulli eqn.** also gives a stagnation pressure (static + dynamic pressure)

$$p_o = p + \frac{1}{2} \rho v^2$$

- Expand compressible p_o in **Taylor series**

$$(1+x)^n = 1 + nx + \frac{n(n-1)}{2} x^2 + \dots$$

$$\frac{p_o}{p} = \left(1 + \frac{\gamma-1}{2} M^2\right)^{\frac{\gamma}{\gamma-1}} = 1 + \frac{\gamma}{\gamma-1} \frac{\gamma-1}{2} M^2 + \frac{\gamma}{2(\gamma-1)} \left(\frac{\gamma}{\gamma-1} - 1\right) \left(\frac{\gamma-1}{2} M^2\right)^2 + \dots$$

$$\frac{p_o}{p} = 1 + \frac{\gamma}{2} M^2 + \frac{\gamma}{2} \left(\frac{M^2}{2}\right)^2 + \dots = 1 + \frac{\rho v^2}{2p} + \frac{\rho v^2}{8p} M^2 + \dots$$

use $M^2 = \frac{v^2}{\gamma p / \rho}$

$$p_o = p + \frac{1}{2} \rho v^2 + \frac{1}{2} \rho v^2 \frac{M^2}{4} + \dots$$

Bernoulli higher terms negligible for small M (<0.3) $0.3^2/4=0.0225$

Stagnation Density and Tables

- **Stagnation Density**

- from T_o , p_o and ideal gas law ($\rho=p/RT$)
- ρ_o **constant** for **isentropic** flow

$$\frac{\rho_o}{\rho} = \left(\frac{T_o}{T} \right)^{1/\gamma-1}$$

$$\frac{\rho_o}{\rho} = \left(1 + \frac{\gamma-1}{2} M^2 \right)^{1/\gamma-1} \quad \text{(VI.8)}$$

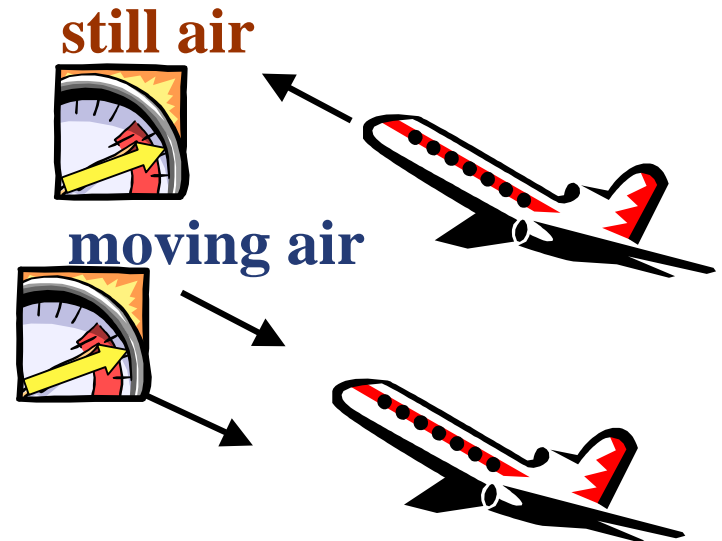
- **Tables**

- T_o/T and p_o/p tabulated in text (John) as function of M listed as T_t/T and p_t/p (t for *total* T and total p)
- $\gamma=5/3$ (Table A.3): atoms (Ar, He, ...) at “not too high” T
- $\gamma=1.4$ (Table A.1): diatomics (N_2 , O_2 , ...) at “moderate” T
- $\gamma=1.3$ (Table A.2): more atoms or higher T
- make your own?

Stagnation versus Static Properties

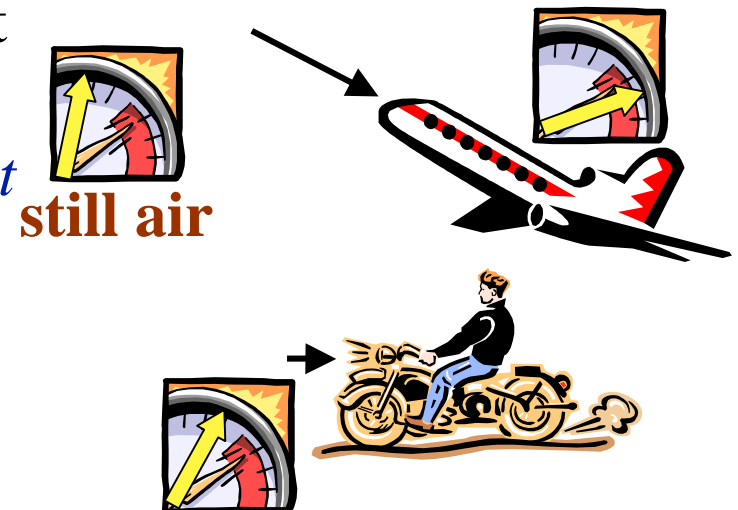
• Static Properties

- represent the properties you would measure if you were moving with the flow (at the local flow velocity)
- always defined in the flow's reference frame



• Stagnation Properties

- always defined by conditions at a point
- represent the (static) properties you'd measure if you first brought the fluid *at that point to a stop* (**isentropically**) *with respect to a chosen observer*
- depends on observer's reference frame

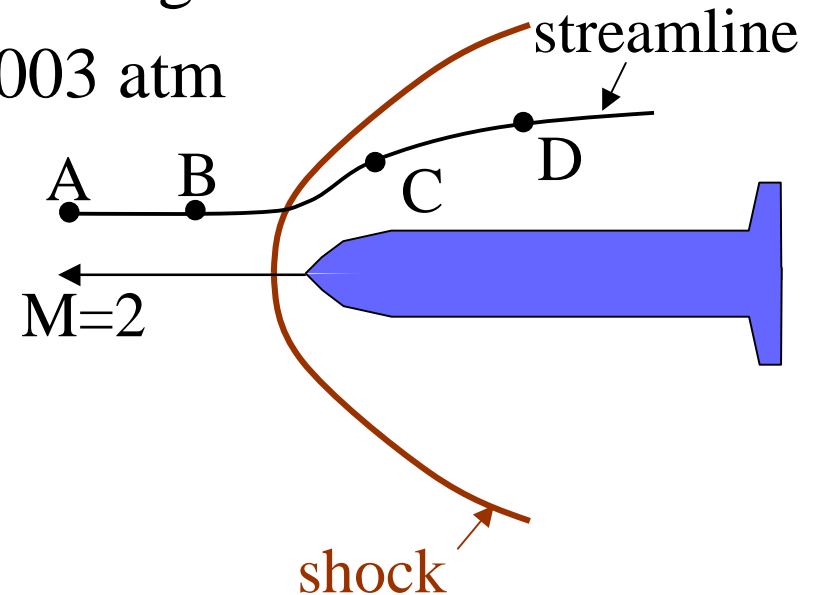


Stagnation Properties: Example

- **Supersonic** projectile ($M=2$) flying through still air
- **Static** conditions: $T_\infty=250\text{ K}$, $p_\infty=0.003\text{ atm}$

$$T_\infty=250\text{K}$$

$$p_\infty=0.003\text{atm}$$



- **Find:**

1. T_o at A (T_{oA}) relative to observer on projectile

2. T_{oD} (same observer) $<$, $>$, $= T_{oA}$?

3. p_{oB} (same observer)

4. p_{oC} (same observer) $<$, $>$, $= p_{oB}$?

Stagnation Properties: Example 2

- Projectile flying through still air at 170 m/s
- Static conditions: $T_\infty = 288$ K, $p_\infty = 1$ atm
- Nose of projectile = point B
- Find:

1. p_{oA} (relative to observer on projectile)

2. p_B

3. T_B

- Hint, use $a = \sqrt{\gamma RT}$

