

## Chemical Equil: Number of $K_p$ Eqns.

- How many equilibrium stoichiometric reactions ( $K_p$ 's) are required to determine equilibrium composition of chosen set of species?
- Assume mixture of  $M$  species having  $\alpha$  kinds of atoms (nuclei)
- Generally  $M+2$  unknowns (2 intensive TD props.)
- IF we specify the 2 TD properties (e.g,  $T, p$  or  $h, p$ ) AND the number/ratio of atomic nuclei
  - that leaves  **$M-\alpha$  unknowns** which requires  **$M-\alpha$  independent  $K_p$  expressions** (=number of reaction degress of freedom)

## Examples

- Mixture:  $H_2$  and  $O_2$ 
  - how many  $K_p$  required? *Red colored expressions*
  - $M=2, \alpha=2 \Rightarrow M-\alpha=0$   $K_p$  required *- are formation reactions*
- Mixture:  $H_2, O_2, H_2O$  e.g.,  $H_2 + \frac{1}{2} O_2 \leftrightarrow H_2O$ 
  - how many  $K_p$  required? or  $2H_2O \leftrightarrow 2H_2 + O_2$
  - $M=3, \alpha=2 \Rightarrow M-\alpha=1$   $K_p$  required
- Mixture:  $H_2, O_2, H_2O, OH, O, H$  e.g.,  $H_2 + \frac{1}{2} O_2 \leftrightarrow H_2O$  (1)  
 $\frac{1}{2} H_2 \leftrightarrow H$  (2)  
 $\frac{1}{2} O_2 \leftrightarrow O$  (3)  
 $\frac{1}{2} H_2 + \frac{1}{2} O_2 \leftrightarrow OH$  (4)
  - how many  $K_p$  required?
  - $M=6, \alpha=2 \Rightarrow M-\alpha=4$   $K_p$  required

*could replace (2), (3) or (4) or?  $H+O \leftrightarrow OH$*

*could replace (1), (2) or (4) or?  $H+OH \leftrightarrow H_2O$*

## Examples (con't)

- Mixture of  $\text{NH}_3$ ,  $\text{HCl}$ ,  $\text{NH}_4\text{Cl}$ 
  - how many  $K_p$  required?
  - $M=3, \alpha=3 \Rightarrow M-\alpha=0$   $K_p$  required ??
- No!
  - for this problem, only 2 of the atom conservation equations are independent

$$\begin{array}{rcl}
 \text{N: } & dn_{\text{NH}_3} + dn_{\text{NH}_4\text{Cl}} & = 0 & 3\times \\
 \text{Cl: } & dn_{\text{HCl}} + dn_{\text{NH}_4\text{Cl}} & = 0 & + 1\times \\
 \text{H: } & 3dn_{\text{NH}_3} + dn_{\text{HCl}} + 4dn_{\text{NH}_4\text{Cl}} & = 0 & =
 \end{array}$$

*Number of independent equilibrium expressions required is  $M-\alpha^*$   
where  $\alpha^*$  is number of INDEPENDENT atom conservation equations*

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## Number of Independent Reactions

- How do you determine how many of the atom conservation equations are independent?
  - linear algebra can be used
  - atomic formation reaction approach
    - write a reaction forming each non-monatomic species from its atoms
    - combine equations to eliminate any atom not actually a species in chosen composition

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