

Energy Level Population Distributions

- Revisit how molecules (e.g., diatomic) are distributed among energy levels in equilibrium

- Boltzmann fraction $\frac{N_i}{N} = f_i(T) = \frac{g_i e^{-\epsilon_i/kT}}{Q}$

- Assuming energy modes separable

$$\begin{aligned} \frac{N(n_{el,i}, v, J, n_x, n_y, n_z)}{N} &= \frac{g_{el,i} g_v g_J e^{-(\epsilon_{el,i} + \epsilon_v + \epsilon_J + \epsilon_{n_x, n_y, n_z})/kT}}{Q_{el} Q_{vib} Q_{rot} Q_{tr}} \\ &= \frac{g_{el,i} e^{-\epsilon_{el,i}/kT}}{Q_{el}} \frac{g_v e^{-\epsilon_v/kT}}{Q_{vib}} \frac{g_J e^{-\epsilon_J/kT}}{Q_{rot}} \frac{e^{-\epsilon_{n_x, n_y, n_z}/kT}}{Q_{tr}} \\ &= f_{el}(n_{el,i}) f_{vib}(v) f_{rot}(J) f_{tr}(n_x, n_y, n_z) \end{aligned}$$

Energy Level Distributions-1
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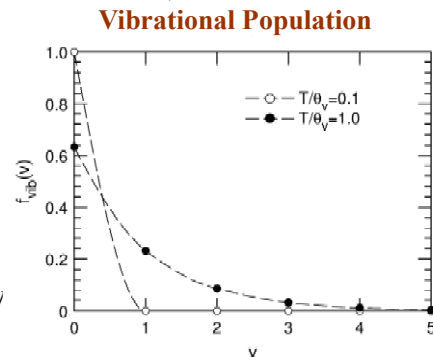
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Internal Energy Mode Populations

- Electronic
 - already examined
- Vibration (single harmonic-oscillator)

$$\begin{aligned} f_{vib}(v) &= \frac{e^{-\theta_v(v+1/2)/T}}{e^{-\theta_v/2T} / (1 - e^{-\theta_v/T})} \\ &= e^{-v\theta_v/T} (1 - e^{-\theta_v/T}) \end{aligned}$$

- $v=0$ always largest fraction
- f_{vib} drops monotonically with v



Energy Level Distributions-2
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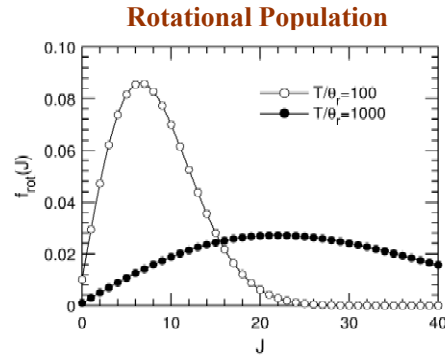
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Internal Energy Mode Populations

- Rotation ($T \gg \theta_r$)

$$f_{rot}(J) = \frac{(2J+1)e^{-\theta_r J(J+1)/T}}{T/\sigma\theta_r}$$

- Why does f_{rot} have peak, while f_{vib} always maximum for $v=0$?



Energy Level Distributions-3

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Translational Energy Mode Population

- Translation

$$f_{tr}(n^2) = \frac{e^{-\theta_{tr} n^2/T}}{(2\pi m k T / h^2)^{3/2} V}$$

– with

$$n^2 = n_x^2 + n_y^2 + n_z^2$$

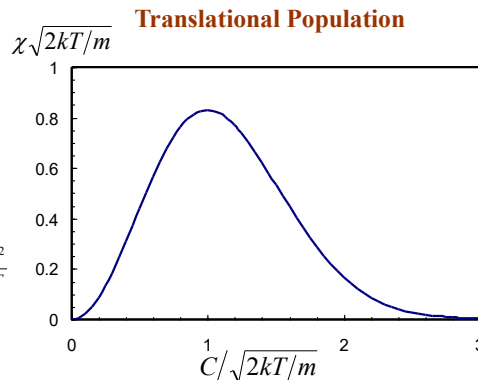
$$\theta_{tr} = h^2 / (8mV^{2/3})$$

– or in terms of speed

$$C = \sqrt{2\varepsilon_{tr}/m}$$

$$\chi(C) = 4\pi \left(\frac{m}{2\pi k T} \right)^{3/2} C^2 e^{-\frac{mC^2}{2kT}}$$

Maxwell (-Boltzmann)
Speed Distribution



Energy Level Distributions-4

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Probability Distributions

- Given these probability distributions, one can ask questions such as:
 - which energy level is most probable at given T

$$\frac{\partial}{\partial i} f_i = 0$$
 - which energy level provides greatest contribution to energy of mode at given T

$$\frac{\partial}{\partial i} (f_i \varepsilon_i) = 0$$
 - at what temperature does a given level i reach a maximum value

$$\frac{\partial}{\partial T} f_i = 0$$

Boltzmann Plots

- Recall what happens if we plot Boltzmann fractions

$$f_i(T) = \frac{g_i e^{-\varepsilon_i/kT}}{Q}$$

$$\Rightarrow \ln \frac{f_i}{g_i} = -\frac{\varepsilon_i}{kT} - \ln Q$$
 - a way to FIND temperature
- Could make such a plot for each energy mode and if each internally in equilibrium
 - get T_{tr} , T_{rot} , T_{vib} , T_{el}
 - if system in equilibrium,
 - if modes not in equilibrium with each other,
- Also possible that distribution not Boltzmann

