

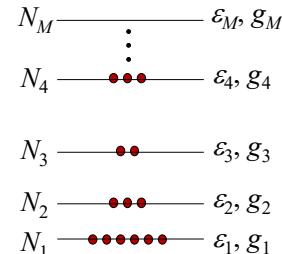
## Review Macrostates

- So far we have found the number of microstates  $W(N_i)$  in a *given* macrostate, e.g.,

$$W_{Boltzmann}(N_i) \approx \prod_{i=1}^M g_i^{N_i} / \prod_{i=1}^M N_i!$$

- macrostate  $\equiv$  given  $N_i$  distribution
- $N_i/N \equiv$  fraction of molecules in  $i^{\text{th}}$  energy level ( $N = \text{total \# molec.}$ )

- Each “**allowed**” macrostate must meet overall constraints  $\sum N_i = N; \sum N_i \varepsilon_i = E$



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## Most Probable Macrostate

- Recall one of our goals is to find total number of possible microstates  $\Omega = \sum_{\text{allowed macrostates}} W(N_i)$
- Turns out (show later) that for  $N$  large  $\ln \Omega \cong \ln W_{\max}$
- Also having no reason not too, we will postulate that *all microstates are equally probable principle of equal a priori probability* so probability of system being in given macrostate  $= W/\Omega$ 
  - thus macrostate with  $W_{\max}$  is also the **most probable macrostate** (at equilibrium)
  - and  $N_i/N$  for the most probable macrostate will tell us the probability of finding a molecule (particle) in a given energy level (at equilibrium)

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## Distribution of Microstates

- To demonstrate  $W_{\max}$  versus  $\Omega$ , consider putting  $N$  particles in  $M$  large boxes, each with  $g$  little boxes
  - for this starting simple demonstration, we are **ignoring energy issues** (simplifies calculations)
- First ask, **how many “macrostates”?**  
(different  $N_i$  distributions)
  - how many ways to put  $N$  things in  $M$  big boxes with no limit on  $N$  per big box
    - same question as B-E statistics
  - example:  $N=6, M=3, g=24$

$$n_{Macro} = \frac{(N+M-1)!}{N!(M-1)!} = \frac{8!}{6!2!} = 28$$

$M=3$	
$N=6$	
•	
•	
•	
•	
•	
$g=24$	
1 = (6,0,0)	
2 = (5,1,0)	
3 = (4,2,0)	
4 = (3,3,0)	
5 = (2,4,0)	
6 = (1,5,0)	
7 = (0,6,0)	
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$\Sigma=28$	✓

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## Simplified Example: Comparing $W$

- What is  $W$  for a given “macrostate”?
  - assuming limit of one per little box (F-D stats.)
  - $W(N_i) = \prod \frac{g!}{N_i!(g-N_i)!}$
  - for our example, some cases
    - all in one  $i$
    - nearly uniform
    - uniform

$$W_{(6,0,0)} = W_{(0,6,0)} = W_{(0,0,6)} = 1.3 \times 10^5$$

$$W_{(3,2,1)} = W_{(3,1,2)} = W_{(2,3,1)} = 1.3 \times 10^7$$

$$W_{(1,3,2)} = W_{(1,2,3)} = W_{(2,1,3)} = 2.1 \times 10^7 = W_{\max}$$

for our equal  $g$ 's (*and no energy issues*)  
 ⇒ uniform distribution most probable ( $W_{\max}$ )  
 ⇒ nearly uniform distributions very likely

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## Simplified Example: # Microstates

- How many total “microstates”?

– how many ways to put  $N$  things in  $M \times g$  little boxes with limit of 1 per little box

– same as F-D statistics

$$\Omega = \frac{(Mg)!}{N!(Mg-N)!} = \frac{72!}{6!64!} = 1.56 \times 10^8$$

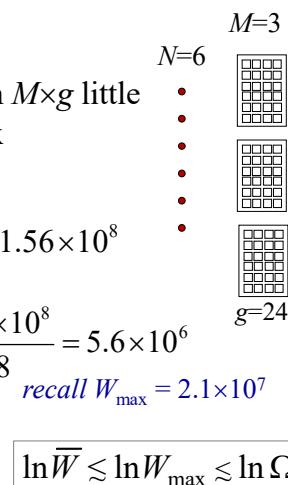
- What is mean  $\bar{W}$ ?

$$\bar{W} = \frac{\Omega}{n_{Macro}} = \frac{1.56 \times 10^8}{28} = 5.6 \times 10^6$$

recall  $W_{\max} = 2.1 \times 10^7$

- Compare  $\ln \Omega$ ,  $\ln W_{\max}$  and  $\ln \bar{W}$

$$\ln \bar{W} \approx 15.5; \ln W_{\max} \approx 16.9; \ln \Omega \approx 18.9$$



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## What Happens for Large $N$

- For very large numbers, these values get closer

– e.g.,  $N=10^{19}$ ,  $M=8000$ ,  $g=1.25 \times 10^{20}$

$$\ln W_{\max} = 1.25129 \dots \times 10^{20}; \ln \Omega = 1.25129 \dots \times 10^{20} \Rightarrow \ln W_{\max} \cong \ln \Omega$$

- Physical interpretation of this example

– 1 cm<sup>3</sup> region at SATP,  $N \sim 10^{19}$

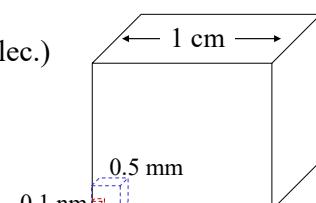
– 1 Å<sup>3</sup> small boxes (about size of molec.)  
⇒  $10^{24}$  possible locations (“states”)

– how many  $N_i$  in larger 0.5 mm “big boxes” ( $M = 8000$ )

• probability of all molecules in 1 “big” box  $\sim e^{-10^{20}}$

• probability of exactly same # molec in each “big” box (most probable “macrostate”)  $\sim e^{-10^{10}}$

• probability of at least  $1.25 \times 10^{15}$  molec in each “big” box (nearly same, uniform distribution)  $\sim 1$



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## Most Probable Energy Macrostate

- So for most macroscopic (large) systems, we can use
$$\ln W_{\max} \cong \ln \Omega$$
  - we can focus on  $\ln W_{\max}$  if we want to find  $\ln \Omega$
  - note: not saying  $\Omega = W_{\max}$ , just that  $\ln$  of these values is essentially the same
- Most probable distribution of particles over energy levels
  - contains nearly all the microstates
  - has nearly 100% probability
  - **represents the TD equilibrium particle distribution**
    - e.g., if we made a measurement of an equilibrium system, it is the distribution we would find 99.999...% of the time