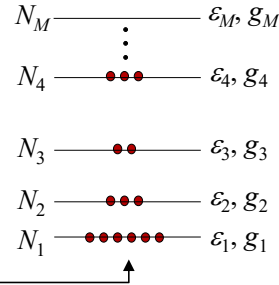


Review Macrostates

- So far we have found the number of microstates $W(N_i)$ in a *given* macrostate, e.g.,

$$W_{\text{Boltzmann}}(N_i) \approx \prod_{i=1}^M g_i^{N_i} / \prod_{i=1}^M N_i!$$

- macrostate \equiv given N_i distribution
- $N_i/N \equiv$ fraction of molecules in i^{th} energy level ($N = \text{total \# molec.}$)



- Each “**allowed**” macrostate must meet overall constraints $\sum N_i = N$; $\sum N_i \epsilon_i = E$

Most Probable Macrostate-1
Copyright © 2009, 2022, 2023 by Jerry M. Seltzman.
All rights reserved.

AE/ME 6765

Most Probable Macrostate

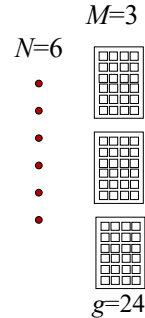
- Recall one of our goals is to find total number of possible microstates $\Omega = \sum_{\text{allowed macrostates}} W(N_i)$
- Turns out (show later) that for N large $\ln \Omega \cong \ln W_{\text{max}}$
- Also having no reason not to, we will postulate that *all microstates are equally probable* ↑ for the macrostate with the most microstates
principle of equal a priori probability
so probability of system being in given macrostate = W/Ω
 - thus macrostate with W_{max} is also the **most probable macrostate** (at equilibrium)
 - and N_i/N for the most probable macrostate will tell us the probability of finding a molecule (particle) in a given energy level (at equilibrium)

Most Probable Macrostate-2
Copyright © 2009, 2022, 2023 by Jerry M. Seltzman.
All rights reserved.

AE/ME 6765

Distribution of Microstates

- To demonstrate W_{\max} versus Ω , consider putting N particles in M large boxes, each with g little boxes
 - for this starting simple demonstration, we are **ignoring energy issues** (simplifies calculations)
- First ask, **how many “macrostates”?** (different N_i distributions)
 - how many ways to put N things in M big boxes with no limit on N per big box
 - same question as B-E statistics
 - example: $N=6, M=3, g_i=24$



$$n_{\text{Macro}} = \frac{(N + M - 1)!}{N!(M - 1)!} = \frac{8!}{6!2!} = 28$$

1 = (6,0,0)
 2 = (5,1,0) (5,0,1)
 3 = (4,2,0) (4,1,1) (4,0,2)
 4 = (3,3,0) (3,2,1) (3,1,2) (3,0,3)
 5 = (2,4,0) (2,3,1) (2,2,2) (2,1,3) (2,0,4)
 6 = (1,5,0) (1,4,1) (1,3,2) (1,2,3) (1,1,4) (1,0,5)
 7 = (0,6,0) (0,5,1) (0,4,2) (0,3,3) (0,2,4) (0,1,5) (0,0,6)

 $\Sigma=28$ ✓

Most Probable Macrostate-3
Copyright © 2009, 2022, 2023 by Jerry M. Seltman.
All rights reserved.

AE/ME 6765

Simplified Example: Comparing W

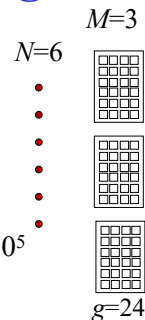
- What is W for a given “macrostate”?
 - assuming limit of one per little box (F-D stats.)

$$W(N_i) = \prod \frac{g!}{N_i!(g - N_i)!}$$
 - for our example, some cases
 - all in one i

$$W_{(6,0,0)} = W_{(0,6,0)} = W_{(0,0,6)} = 1.3 \times 10^5$$
 - nearly uniform

$$W_{(3,2,1)} = W_{(3,1,2)} = W_{(2,3,1)} = W_{(1,3,2)} = W_{(1,2,3)} = W_{(2,1,3)} = 1.3 \times 10^7$$
 - uniform

$$W_{(2,2,2)} = 2.1 \times 10^7 = W_{\max}$$
- for our equal g 's (and no energy issues)
 \Rightarrow uniform distribution most probable (W_{\max})
 \Rightarrow nearly uniform distributions very likely



Most Probable Macrostate-4
Copyright © 2009, 2022, 2023 by Jerry M. Seltman.
All rights reserved.

AE/ME 6765

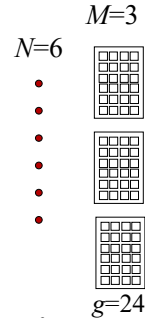
Simplified Example: # Microstates

- How many total “microstates”?
 - how many ways to put N things in $M \times g$ little boxes with limit of 1 per little box
 - same as F-D statistics
- $$\Omega = \frac{(Mg)!}{N!(Mg - N)!} = \frac{72!}{6!64!} = 1.56 \times 10^8$$
- What is mean \bar{W} ?

$$\bar{W} = \frac{\Omega}{n_{\text{Macro}}} = \frac{1.56 \times 10^8}{28} = 5.6 \times 10^6$$
 - Compare $\ln \Omega$, $\ln W_{\text{max}}$ and $\ln \bar{W}$

$\ln \bar{W} \approx 15.5$; $\ln W_{\text{max}} \approx 16.9$; $\ln \Omega \approx 18.9$

$\ln \bar{W} \lesssim \ln W_{\text{max}} \lesssim \ln \Omega$



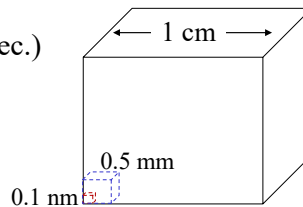
Most Probable Macrostate-5
Copyright © 2009, 2022, 2023 by Jerry M. Seltman.
All rights reserved.

AE/ME 6765

What Happens for Large N

- For very large numbers, these values get closer
 - e.g., $N=10^{19}$, $M=8000$, $g=1.25 \times 10^{20}$

$\ln W_{\text{max}} = 1.25129... \times 10^{20}$; $\ln \Omega = 1.25129... \times 10^{20} \Rightarrow \ln W_{\text{max}} \cong \ln \Omega$
- Physical interpretation of this example
 - 1 cm³ region at SATP, $N \sim 10^{19}$
 - 1 Å³ small boxes (about size of molec.)
 $\Rightarrow 10^{24}$ possible locations (“states”)
 - how many N_i in larger 0.5 mm
“big boxes” ($M = 8000$)
 - probability of all molecules in 1 “big” box $\sim e^{-10^{20}}$
 - probability of exactly same # molec in each “big” box (most probable “macrostate”) $\sim e^{-10^{10}}$
 - probability of at least 1.25×10^{15} molec in each “big” box (nearly same, uniform distribution) ~ 1



Most Probable Macrostate-6
Copyright © 2009, 2022, 2023 by Jerry M. Seltman.
All rights reserved.

AE/ME 6765

Most Probable Energy Macrostate

- So for most macroscopic (large) systems, we can use

$$\ln W_{\max} \cong \ln \Omega$$

- we can focus on $\ln W_{\max}$ if we want to find $\ln \Omega$
- note: not saying $\Omega = W_{\max}$, just that \ln of these values is essentially the same
- Most probable distribution of particles over energy levels
 - contains nearly all the microstates
 - has nearly 100% probability
 - **represents the TD equilibrium particle distribution**
 - e.g., if we made a measurement of an equilibrium system, it is the distribution we would find 99.999...% of the time